Chapter 8 Microbial Metabolism: The Chemical Crossroads of Life

Building Your Knowledge

1) What are the two branches of metabolism?

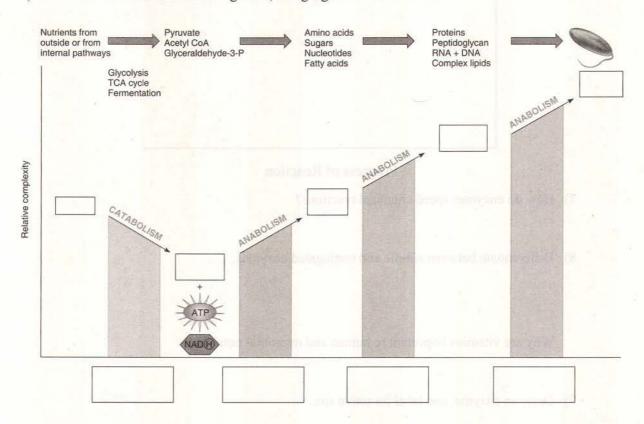
a. _____

b.

Which branch synthesizes large molecules from small subunits?

Which breaks down large molecules into small subunits?

2) Fill in the boxes on the following table, using figure 8.1 as a reference.



- 3) What do catalysts do in a chemical reaction?
- 4) Do enzymes add energy to chemical reactions?

Are they changed by the reaction?

Do they interact with several substrate molecules or one molecule per enzyme (then the enzyme goes away)?

5) What a	re enzymes made up of—proteins, lipids, or sugars?
	simple reaction on the following graph. Label the products, reactants, and energy of on (Ea).
	to in the second of the second
Energy	complete the compl
	Progress of Reaction
7) How do	enzymes speed chemical reactions?
8) Differe	ntiate between simple and conjugated enzymes.
Why ar	e vitamins important to human and microbial nutrition?
9) Draw a	n enzyme, and label its active site.
10) Endoen	zymes work inside the cell. What are enzymes that work outside a cell
called?	
11) Enzymo	es that are present all the time are called
	Security of all negligible regarded
Induced	d enzymes are activated or produced only when is present.

If a molecule is reduced, d	oes it gain or lose elec	etrons?	
If a molecule is oxidized, o	loes it gain or lose elec	ctrons?	
13) How do enzymes contribut Pseudomonas aeruginosa,			voccus pyogenes,
Organism	Enzyme	Diseas	se
Streptococcus pyogenes,			
Pseudomonas aeruginosa			
Clostriduium perfringes	li(n) s	11 12 2	
14) Differentiate between com15) What is energy?	petitive and noncompe	titive inhibition of e	nzyme activity.
Describe 3 forms of energy	/,		
Which forms of energy are	most commonly used	in cells?	
16) How do endergonic and ex	ergonic reactions diffe	er?	
Which are typically anabol	ic?		
Which are typically catabo	lic?		
17) What is ATP and why is it	called "metabolic mor	ney"?	

12) How are oxidation and reduction related?

19) Which yields more energy, anaerobic respiration or aerobic respiration?
Which requires oxygen?
20) What is the basic equation for aerobic respiration in microbes?
For every glucose molecule burned, the cell needs oxygen molecules, and produces molecules of carbon dioxide and molecules of water.
21) What is the final electron acceptor in aerobic respiration?
22) Does the TCA cycle reduce or oxidize glucose?
23) Glycolysis starts with and ends with
How many ATP molecules are generated in glycolysis for each molecule of glucose consumed
24) How many carbons are in a glucose molecule?
How many carbons are in a pyruvic acid molecule?
How many pyruvic acid molecules are produced for every glucose molecule metabolized?
25) The TCA cycle produces and
Where do NADH molecules go with their electrons?
26) Which stage of glucose metabolism requires a membrane?
Why?

18) What are the three coupled pathways of catabolism?

28) Draw an ATP synthase molecule, a membrane, the H+ gradient, the flow of H+ ions and the formation of ATP from ADP and P.
29) Where and how is water formed during aerobic respiration?
30) Why do we consider pyruvic acid a central part of metabolism?
What can pyruvate be converted to anaerobically?
31) Which form of glucose metabolism yields more energy—anaerobic or aerobic?
Where is most ATP generated?
32) How do fermentation and anaerobic respiration differ?
Which yields more energy per glucose molecule?
33) How do alcoholic and acidic fermentation differ?
Which fermentation do you want if you are making bread or beer?
Which process sours milk?
Which process do you want if you are making yogurt?
What happens when you work out to the point when your muscles are deprived of oxygen?

27) How does ATP synthase generate ATP?

What do amino ac	ids combine to	form?			
35) How are carbohyd					
Where are carbohy	ydrates used in	a bacterial cell?	<u>, </u>		
36) How are lipids (fa	ts) made?				
What are they used	d for in a proca	ryotic cell?			
37) Many metabolic p	athways are an	nphibolic. What	t does this mean?		
28) Do maooyaan mal			ide) some from the		aut abain (Vas
38) Do precursor mole or No)?	ecules (amino a	icids, sugars, np	ids) come from the	electron transp	ort chain () es
Where may they c	ome from?				
39) If we labeled a gluwould the carbons			radioactively, so the	ey could be tra	ced, where
			er of the said grant set		

34) Amino acids are made up of carbon and nitrogen. Where can cells get the carbon?

Organizing Your Knowledge

Part of Aerobic Respiration	Location	Starting Molecules	End Products
glycolysis			
	The state of the s	1 1 1 1 1 2	
TCA cycle	4 44 44		
Electron transport chain			
Election transport chain			

Metabolic Mechanism	Pathways Included	Final Electron Acceptor	Products	Microbes Using This
Aerobic respiration				
Anaerobic fermentation				
Anaerobic respiration				

Microbial Metabolism

PWKUEJYCWDYZTZFQXAMEQROKRRGRNQ LDOKGYYXTPSIGITYTGVIAOMMBBNKAA RQZUMBBWLKPITNCWCQYSANXFQWNABI CZNBNEFZWQJZLGZSDCYMXRKXKUJDQL ZOYSCRVSLGDYPTRDUZKGEUKWEOYLHX PWHZDCWGLYCOLYS I SBXCCTFH I SBEBF MOLENACONSTITUTIVESMXNAPKQQDOH RTKJOHMFXFWALOZKHLUTAUPBEHKWUL V L DWC G T D R K C B C U Y F E O Z E R U E O O R A H T E BKHQTGEJEIVPATCKNDMERADNXLYNAC FXXRMYFPWRIDTDXXJADFWETRJYIZZH AQZKJKPWKDAMDJEWYGYQKXDEFNTSUZ HJEPQMECOMPETITIVEINHIBITIONMW ZYWOUHNTIJHBAMEVEMKLFIZGNEPKEA ZODPOEEABCTEVJOZSVNWESNRCAPFVW BVZRQKRBGLAFCEBVBSKSRQCDAGCCWA MQHIOZGWTWUKQFKFSTTFMOSKUYNQQK ULUSVLYLFVYDBDVYORTIEOJDICXROD J C P G H N Y O F Z P J E D I S E N T X N C Q O Z E E W H X Y S B N T R F S W E J B J V S P G X J F T B H P C P C D R O CUNRCPWHIVUWTGTQAFYKABAJQADTNW ANABOLISMSYLPOLUMDWXTRRATOIWKF LYTCACYCLEHJKOLEYRVBIJXAENZYME GADWXQHMFYPIDFPEZJVHOQBAFILCVD GVBXLNXISWNWYTDRFLCZNOVVSUKHXN UJGIWOCCLAAMPHIBOLICLCOFAC VXAZLDHHSREDUCTIONFIRSENVO TIONFIRSENVOEBAS IOHLEEEWNOMVYEGWDSQOENOEKOTF SNXENDERGONICZTOOMPLLWYFOGPFLO OOULOSNZQBQVPIVFLFTOZJKMGJLPGS

- (___) Energy currency of a cell ___) Cycle that takes in pyruvic acid and converts it to CO2 and provides NADH for the electron transport chain
-) Metabolic pathways that can be used for anabolism and catabolism
- (______) Synthesis of large molecules from small ones (______) Breakdown of
- large molecules into small ones
 (_____) Metallic ions
 associated with enzymes that are critical to enzyme function
-) Process by which a substance binds to the active site of an enzyme and stops it from binding to its substrate
- _) Enzyme that is present in constant concentrations, independent of substrate concentration

-) Chemical reaction that requires the addition of
- __) The ability to do w ork __) Protein catalyst that speeds reactions by low ering the energy of activation
-) Anaerobic process that leads to the production of gases, acids and/or alcohol
- 13. (______) Pathw ay that converts glucose to pyruvic acid
 14. (______) The addition of w ater to break bonds
- 15. (_____) Enzyme that is found in higher concentrations when its substrate is present
- 16. (____) Molecules that are chemically unstable are called
- _) Sum of all chemical and physical activities converting energy to usable forms AND using energy to do work

- _) The gaining of electrons
-) Product of Streptococcus pyogenes that dissolves blood clots
- 20. () Molecule that enzymes interact with to produce products

Practicing Your Knowledge	has opened. Printed the Land
1. Enzymes	7. Enzyme cofactors are
a. add energy to chemical reactions b. increase the rate of chemical reactions c. are changed by the chemical reactions they catalyze d. work on all chemical reactions the same way 2. What is the final electron acceptor in aerobic respiration?	a. generally vitamins and used to support enzyme function b. generally apoenzymes and work alone c. generally metallic and activate enzymes d. not used in bacterial cells; procaryotes have simple enzymes
a. oxygen b. carbon dioxide c. sulfur d. NADH	8. If you labeled the carbons of glucose and sent them through aerobic respiration, where and how would the carbons be released?
Which of the following factors will change enzyme function? a. temperature b. pH	 a. in glycolysis as carbon dioxide b. in glycolysis as water c. in the TCA cycle as water d. in the TCA cycle as carbon dioxide
c. substrate concentration d. all of the above	9. Which portion of aerobic respiration requires a membrane to generate energy?
4. An enzyme inhibitor that binds to the site normally used by a substrate and blocks enzyme function is called a a. positive feedback inhibitor b. competitive inhibitor	a. glycolysisb. TCA cyclec. electron transport chaind. fermentation
c. allosteric inhibitor d. enzyme inducer	10. Which part of central metabolism does NOT contribute precursor molecules to anabolic pathways?
5. The energy of activation of a chemical reaction a. increases when enzymes are present b. decreases when enzymes are present c. is not changed by enzymes	a. TCA cycle b. electron transport chain c. glycolysis d. pyruvic acid
6. Beta-galactosidase is not produced by a cell unless its substrate, lactose, is present. It metabolizes lactose inside the cell. We would describe this as a	a. reduction b. oxidization c. condensation d. induction

a. creates : anabolic b. breaks : anabolic c. creates : catabolic d. breaks : catabolic

12. The addition of water to chemical

bonds ____ them and is a ____ reaction.

a. constitutive endoenzyme

b. induced endoenzyme

c. induced exoenzyme d. constitutive exoenzyme

13. Anabolic reactions ____ energy and are used in a cell for reactions.

a. release : synthesisb. use : degradativec. release : degradative

d. use: synthesis

14. ___ is the energy currency of cellular reactions.

a. DNA

b. phosphate

c. ATP

d. AMP

15. Where is most of the energy (ATP) generated during aerobic respiration?

a. glycolysis

b. TCA cycle

c. fermentation

d. electron transport chain